

Experimental and numerical analysis of syngas Mild combustion

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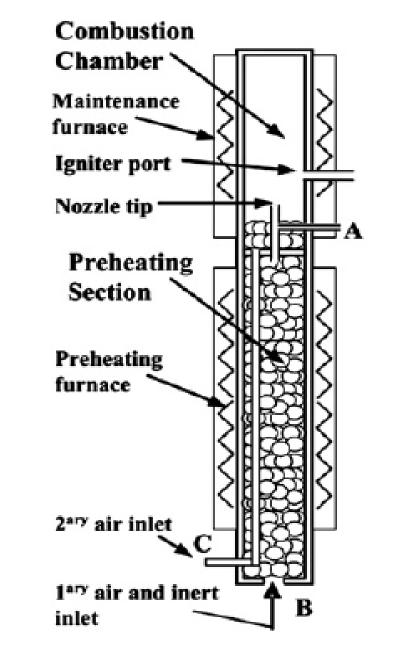
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OBJECTIVE

Aim of this work is to develop a CFD model for prediction of NOx emissions, temperatures and species mole fractions in a Mild combustion experimental burner which was fed with H2 and H2 /CO2/H2O mixtures.

Experimental apparatus



Experimental and modelling runs investigated

Run	H2	Air1	N2	CO2	H2O	Preheater
numbe	(Nml/min)	(Nml/min)	(Nml/min)	(Nml/min)	(Nml/min)	temperature (K)
		(preheated)	(preheated)			
1	407.6	999.5	1293.1	0	0	1173
2	596.6	1463.1	2705.6	149	745	1373
3	596.6	1463.1	543.3	149	745	1373

The models Used in this work

•Turbulence models: Standard K-ε, k-ω-Transition, RSM-Transition.

•Combustion models: Eddy dissipation/Finite Rate (ED/FR), Eddy Dissipation Concept (EDC), modified EDC [5,6].

•Kinetic mechanisms: Dryer-Westbrook mechanism (WD), POLIMI-H2CO1101 [7].

Fuel: H2 /CO2/H2O mixtures

Effect of turbulence models on prediction of middle temperature

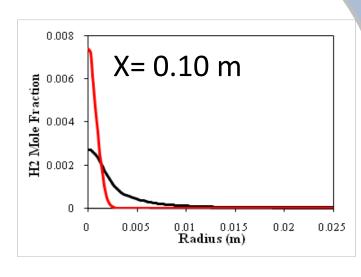
and outlet CO emission (used models: modified EDC with POLIMI-H2CO1101)

CO (ppm, dry basis)		EXP	SKE	k-ω-Transition	RSM-Transition
Run 1	CO (ppm)	0.5	0	0	0
	Temperature (K)	1060	1090	1090	1080
Run 2	CO (ppm)	130	2	42	129
	Temperature (K)	1284	1335	1330	1290
Run 3	CO (ppm)	18	0.7	6	23.2
	Temperature (K)	1285	1300	1294	1280

•The RSM- transition gives a better prediction for temperature and CO.

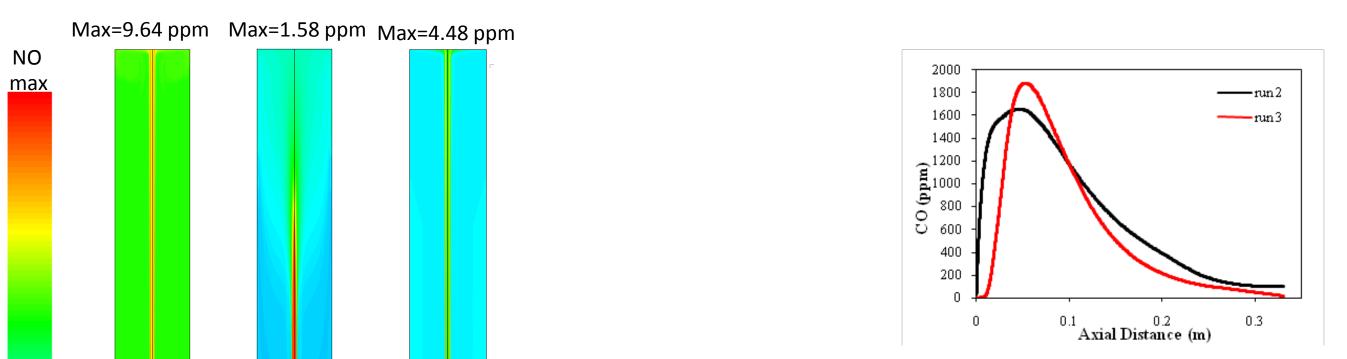
Effect of diffusion on temperature and H2 mole fraction

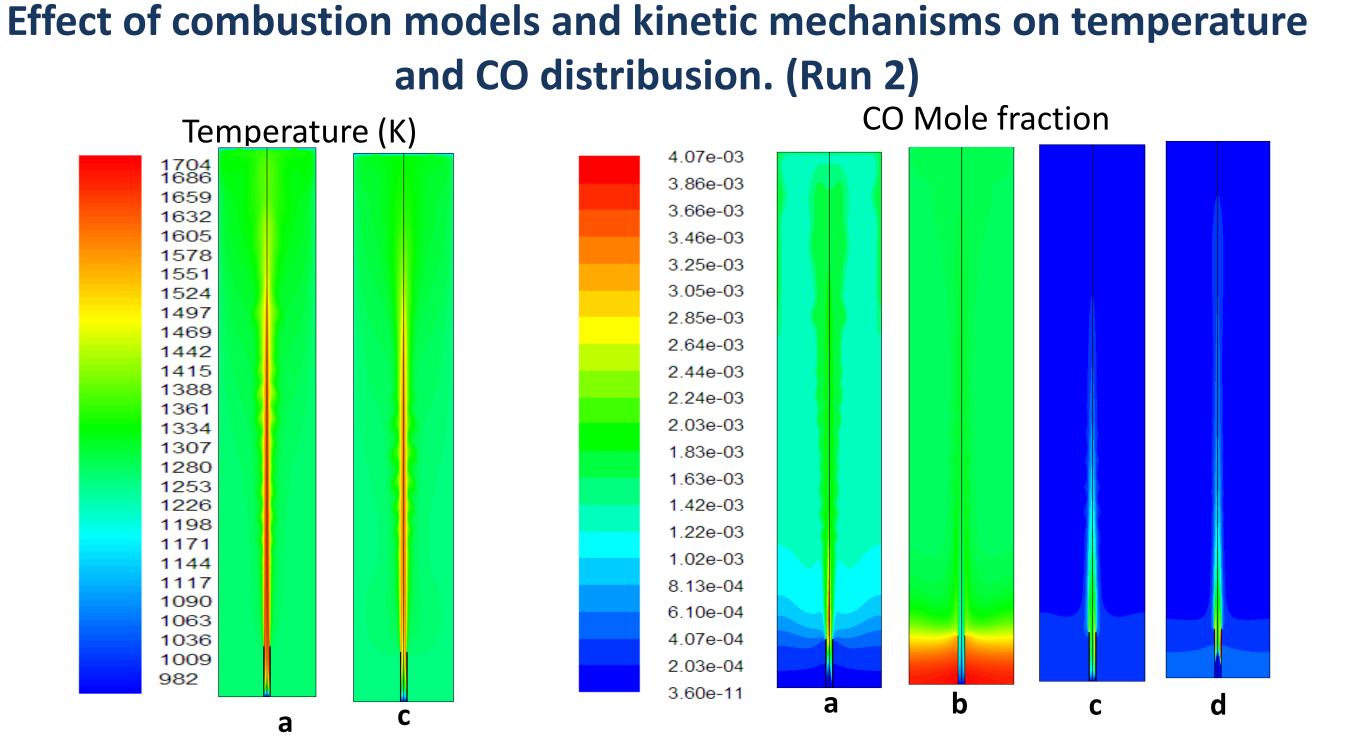
	modified EDC with PC	DLIMI 1101 (Runs 2).
X= 0.05 m	X= 0.10 m	X= 0.05 m



•when laminar diffusion terms are included in the calculations, the temperature profile varies slightly, but the H2 profile varies significantly.

Prediction of CO and NO emissions





(a) ED/FR with (WD) mechanism, (b) modified EDC with (WD) mechanism, (c) modified EDC with POLIMI 1101, (d) EDC with POLIMI 1101.

•Temperature distributions are almost similar for all the combustion models and mechanisms, but the CO distributions are quite different. when the (WD) mechanism is used, the value of CO mole fraction is higher than that of the POLIMI syngas mechanism. •The maximum value of NO belongs to run 1. •a high concentration region of NO can be observed near the central axis of burner in all three runs.

•when the flow rate of N_2 in inlet mixture is increased, the peak value of CO profile at the beginning of chamber decreases.

CONCLUSIONS

The results of this work showed that the KPP and CFD results are in good agreement with experimental observations only when the RSMtransition turbulence model together with modified EDC and POLIMI 1101 syngas mechanism are employed in the model. The results also revealed that as the nitrogen flow rate in the gas mixture increased (runs 2 and 3), the CO mole fraction at the burner outlet increased.

[1] Derudi, M., Villani, A., Rota, R., sustainability of mild combustion of hydrogen-containing hybrid fuels, Proc. Combust. Inst. 31: 3393-3400 (2007) [2] Galletti, C., Parente, A., Derudi, M., Rota, R., Tognotti L., Numerical and experimental analysis of NO emissions from a lab-scale burner fed with hydrogen-enriched fuels and operating in MILD combustion Int. J. Hydrogen Energy 34: 8339-8351 (2009) [3] Parente, A., Galletti, C., Tognotti, L., Effect of the combustion model and kinetic mechanism on the MILD combustion in an industrial burner fed with hydrogen enriched fuels, Int. J. Hydrogen Energy 33: 7553-7564 (2008). [4] Yu, Y., Gaofeng, W., Qizhao, L., Chengbiao, M., Xianjun, X., Flameless combustion for hydrogen containing fuels, Int. J. Hydrogen Energy 35: 2694-2697 (2010) [5] A., De, E., Oldenhof, P., Sathiah, D., Roekaerts, Numerical Simulation of Delft-Jet-in-Hot-Coflow (DJHC) Flames Using the Eddy Dissipation Concept Model for Turbulence–Chemistry Interaction Flow Turbulence Combust. DOI 10.1007/s10494-011-9337-0.

[6] Aminian, J., Galletti, C., Shahhosseini, S., Tognotti, L., Modelling a reacting jet in hot diluted coflow: evaluation of the turbulence/chemistry interaction model, Proc. of the 5th European Combustion Meeting, Cardiff, UK, 2011, pp 1-6.

[7] Frassoldati, A., Faravelli, T., Ranzi, E., Ignition, Combustion and Flame Structure of carbon Monoxide/Hydrogen Mixtures. Note1: Detailed kinetic Modeling of Syngas Combustion also in Presence of Nitrogen Energy 32: 3471-3485 (2007), Available at: http://www.chem.polimi.it/CRECKModeling).

a=run1, b=run2 and c=run3.

